# Modification of ledeburite microstructure on impeller blades by mean of heat treatment

by Rendi Reynaldi

# **General metrics**

<b>14,795</b> characters	<b>2,310</b> words	201 sentences	<b>9 min 14 sec</b> reading time	<b>17 min 46 sec</b> speaking time
Score		Writing	Issues	
54		<b>179</b> Issues left	<mark>92</mark> Critical	<mark>87</mark> Advanced

This text scores better than 54% of all texts checked by Grammarly

## Plagiarism



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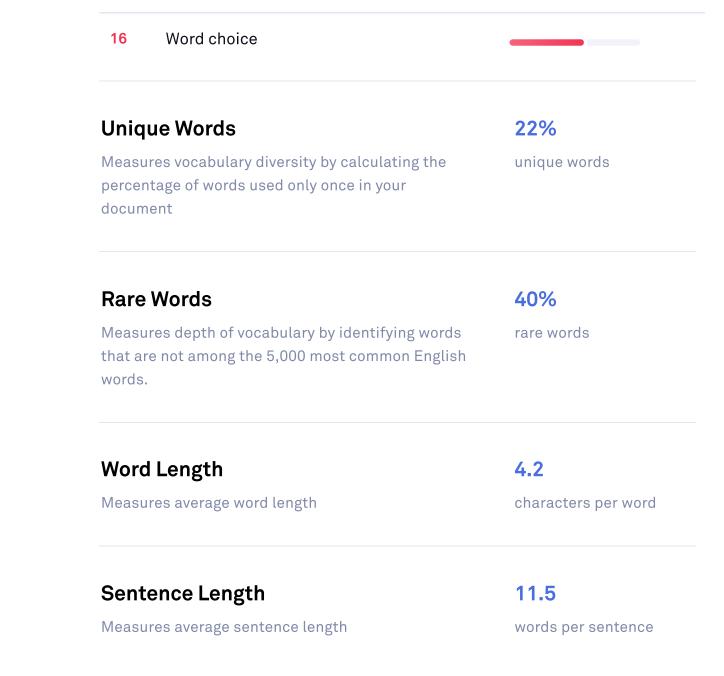


## Writing Issues

116	Correctness	
24	Misspelled words	
28	Unknown words	
11	Determiner use (a/an/the/this, etc.)	
18	Punctuation in compound/complex	
	sentences	
3	Mixed dialects of english	-
16	Improper formatting	
3	Comma misuse within clauses	-
1	Text inconsistencies	•
3	Confused words	-
2	Incomplete sentences	•
1	Incorrect verb forms	•
1	Faulty subject-verb agreement	•
1	Incorrect noun number	•
1	Wrong or missing prepositions	•
1	Misuse of quantifiers	•
1	Pronoun use	•
1	Closing punctuation	•
47	Clarity	
25	Passive voice misuse	
3	Hard-to-read text	-
11	Wordy sentences	
8	Intricate text	
16	Engagement	

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# Modification of ledeburite microstructure on impeller blades by mean of heat treatment

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Modification of ledeburite microstructure on impeller blades by mean of heat treatment

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Abstract. Water pump machine impeller is made of gray cast iron with some variation of wall thickness. The thin part undergoes consequently a very rapid cooling and promotes the formation of hard ledeburite, This study removes the already formed ledeburite  $((\alpha + Fe3C) + Fe3C)$  by mean of heat treatment. By calculating the mass fraction of iron carbide at the level of 5%, heating the material at the austenite temperature, holding it for 2 hours and cooling it subsequently in the air, ledeburite changes to ferrite and graphite. Graphite is formed discrete along the grain boundaries of previous pearlite after the completion of heat treatment. The graphite resembles a randomly oriented eutectic undercooling type D and this is associated with a decrease of hardness to 115HV at the tip of the impeller blade.

## 1. Introduction

The impeller is a part of the water pump machine that serves to produce water pressure. Figure 1 illustrates a water pump image having thin involute blades.

Figure 1. Water pump impeller.

The impeller is made of gray cast iron with a material composition based on Fe with carbon, silicon, manganese alloy and some accompanying elements such as phosphorus and sulphur <sup>10</sup>(table <sup>11</sup>), Grey Cast Iron belongs to the most cost effective material for the impeller and classified as is <sup>14</sup> one of the stable types of cast iron, whose microstructure is characterized by the appearance of graphite, has a carbon content of above 2% and has good casting properties but low elongation value [1].<sup>16</sup>

Appropriate manufacturing processes produce cast iron with microstructures consisting of eutectic lamellar graphite in the matrix of <sup>17</sup> ferrite and pearlite (figure 2a). The most influencing parameters which

180

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2

affect<sup>23</sup> the properties of the material and<sup>24</sup> its microstructures are the elemental compositions and the cooling rate [2].

Since the impeller blade shows variation in its wall thickness, the edge encounters very thin part. In

this section, the metastable solidification takes place and produces a microstructure of  $\frac{^{28}}{[3]}$ 

(figure 2b).

In addition,<sup>29</sup>, <u>the occurrence of</u> <sup>30</sup> this <u>ledeburite</u> <sup>31</sup> can also <u>be initiated</u> <sup>32</sup> by several potential causes, such as high content of carbon, less silicon content (Si) <u>and</u> <sup>33</sup> lack of adequate diffusion time [4].

Table 1. Chemical composition of grey cast iron for several wall thickness [5].

**G** grammarly

Grade
Wall Thickness
(mm)
C
Si
Mn
P max
S
max
FC100
-
3.4-3.9
2.1-2.6
0.5-0.8
0.3
0.15
FC150
<30
3.3-3.5
2.0-2.4
0.5-0.8
0.2
0.12
30-50
3.2-3.5



1.9-2.3			
0.5-0.8			
0.2			
0.12			
>50			
3.2-3.5			
1.8-2.2			
0.6-0.9			
0.2			
0.12			
FC200			
<30			
3.2-3.5			
1.6-2.0			
0.7-0.9			
0.15			
0.12			
30-50			
3.1-3.4			
1.5-1.8			
0.8-1.0			
0.15			
0.12			
>50			



3.0-3.3			
1.4-1.6			
0.8-1.0			
0.15			
0.12			
FC250			
<30			
3.0-3.3			
1.4-1.7			
0.8-1.0			
0.15			
0.12			
30-50			
2.9-3.2			
1.3-1.6			
0.9-1.1			
0.15			
0.12			
>50			
2.8-3.1			
1.2-1.5			
1.0-1.2			
0.15			
0.12			
FC300			



<30			
2.9-3.2			
1.4-1.7			
0.8-1.0			
0.15			
0.10			
30-50			
2.9-3.2			
1.2-1.5			
0.9-1.1			
0.15			
0.10			
>50			
2.8-3.1			
1.1-1.4			
1.0-1.2			
0.15			
0.10			

Reducing the carbon content can considerably be done to avoid the formation of ledeburite, however, this will affect the thicker part of the object, whereby graphite will not be formed.<sup>36</sup> Silicone content is additionally put into the material to prevent white solidification, whereas it causes a decrease of the hardness due to the formation of a larger<sup>38</sup> number of ferrite and bigger<sup>39</sup> size of graphite [6]. Meanwhile, the biggest problem in the manufacturing process of a blade is the ununiform thickness of the blade itself. The microstructure is consequently not uniform. The solidification of metal liquid in the area of smaller wall thickness will form a ledeburite microstructure. <sup>42</sup>The area accordingly becomes very hard, which is also associated with poor elastic and plastic behavior. Due to the hardness and low impact value, this part is not to be machined [7].

Figure 2. The microstructure of FC (A) and <u>ledeburite</u><sup>44</sup> (B), etched with Nital 3%.

Uniform wall thickness and appropriate chemical composition prevent a metastable solidification.

Figure 3. Binary diagram Fe-C (a) and CCT diagram (b).

🔵 grammarly

Occasionally heat treatment process can be done to eliminate the existing ledeburite.<sup>46</sup> Heat-treatment to remove ledeburite is accomplished by heating the sample at the austenisation temperature, as shown in the binary diagram (figure 3a). Cooling rate should be set by using CCT diagram (figure 3b) [8]. By calculating the fraction of ledeburite, calculating the composition/carbon content in the matrix, determining the decomposition temperature, determining the decomposition time and cooling at the apropriate cooling rate. ledeburite can be transformed into Fe3C which is dispersed at the grain boundary [9]. Ledeburite originally presented in the microstructure (figure 4a) can decompose to pearlite and ferrite (figure 4b). The heattreatment caused changes in the shape of pearlite

and graphite size.

The decomposition of Fe3C in the ledeburite <sup>60</sup> is affected by several elements such <sup>61</sup> as manganese <sup>62</sup> (Mn). Manganese impedes the kinetics of graphitization of nodular <sup>63</sup> graphite cast iron [10]. Manganeses stabilizes the cementite and slows down the <sup>64</sup> decomposition kinetics (in iron <sup>65</sup> and graphite).

Figure 4. initial condition with ledeburite (A) and after heat treatment (B), etched with Nital 3%.

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## 2. Design of experiment

2.1. Sampel (100% ledeburite)

The sample used is the tip of the impeller blade which has a 100% ledeburite micro structure. The calculation of the actual amount of C will be used for the formation of a new matrix after the decomposition of Fe3C. The microstructure accordingly has a carbon content of 4.3%.

2.2 Solubility of Fe3C in 5% Ledeburite In this study, the maximum allowable level of <u>ledeburite</u>  $\frac{77}{15}$   $\frac{78}{15}$  up at 5%. Ledeburite consists of a mixture of  $\frac{79}{15}$  phase and cementite (Fe3C). Calculation based on the Fe-Fe3C diagram (figure 3) and lever rule can be applied for dissolving Fe3C in the ledeburite.<sup>81</sup>

Table 2. Percentage of ledeburite former

5% <u>ledeburite</u><sup>83</sup> Microstructure Carbon content Weight percent Austenite (y) 2.0 % 2.47 % Fe3C 6.67 % 2.53 %

Based on the data above it can be concluded that the austenite (y) contained in the material of 5% ledeburite is 2.47%, while the cementite content (Fe3C) is



2.53%.<sup>87</sup>The removal of the ledeburite<sup>88</sup> done by decomposing Fe3C into the matrix.

Table 3. Percentage of carbon in the decomposition of Fe3C.

Microstructure Max.<sup>89</sup> Carbon content

Total carbon content

Fe3C 6.67 % 0.977 %

The design of the heat treatment process is accordingly based <sup>90</sup> on the carbon content of 0.977% C.

2.3. Determining the Heating Temperature
The carbon percentage is used <sup>91</sup>/<sub>91</sub> as the initial reference of temperature
determination for sample heating <sup>92</sup>/<sub>93</sub> it shown on the Fe-Fe3C diagram. <sup>94</sup>/<sub>94</sub> in detail
it can be described as follows: From this Figure <sup>95</sup>/<sub>5</sub>, the
minimum <sup>96</sup>/<sub>10</sub> temperature is at line A1 (727°C) and <sup>97</sup>/<sub>10</sub> critical temperature at Acm line
(± 830 ° C) and <sup>98</sup>/<sub>10</sub> the maximum temperature is below solidus <sup>99</sup>/<sub>10</sub> in (± 1360°C).

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#### <sup>100</sup> □doi:10.1088/1757-899X/541/1/012029

Figure 5. Diagram Fe-Fe3C, decomposition of Fe3C in ledeburite.

The heating is done <u>exactly</u> at the critical temperature of 830°C <u>which</u> is then followed by cooling in the open air.

## 2.4. Holding time

The heating time is set to 2 hours according to the minimum reaction time to disolve the ledeburite for

a thickness of 5mm. The predetermined heating time can dissolve the ledeburite and the carbon from

Fe3C will subsequently diffuse into graphite or be present as Fe3C at the grain boundary.

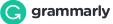
## 2.5. Testing

Tests were performed by applying metallographic analysis using an optical microscope and vickers

hardness testing.

## 3. Result and Analysis

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### 3.1. Initial microstructure

The impeller part having a thick wall (5 mm) shows a microstructure containing graphite with a

dominant matrix of pearlite and a small amount of ferrite (figure 6a). Ferrite itself is formed <sup>110</sup> around the graphite. The graphite that occurs has the shape type of A and random orientation [11]. At the edge of the impeller, the microstructure (figure 4b) shows a dominant fraction of the secondary ledeburite <sup>111</sup> (pearlite + Fe3C) [11]. In some other location <u>assemblies</u> of ferrite around the graphite are still to be found.

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Figure 6. Micrograph at the thick wall (A) and at the tip (B), etched with Nital 3%.

3.2. Microstructure after heat treatment

Metallographic observation after the heat treatment process shows a complete change of the microstructure. Ledeburite decomposes completely into ferrite and graphite (figure 7a) [12]. The microstructure nevertheless shows nearly spherical primary phase, which is already transformed into ferrite. The cementite (Fe3C) is not longer exist since all cementite is transformed into graphite and ferrite. Cementite (Fe3C), which was originally located at the grain boundary, has been transformed into a group of graphite with random orientation. Figure 7b the formation of D-type graphite which is commonly identified as the undercooling eutectic graphite [13]. Conditions without etching.

Figure 7. Micrograph after heat treatment process with graphite tipe D (a) and graphite at the grain boundary with random orientation. (b), without etching.

Figure 8 shows the undissolved (Fe3C) carbide in some particular location of the material, whereby the formation of graphite did not take place. The uneven distribution of manganeses and silicon is considered to be the cause of it [14].

Figure 8. Retained iron carbide, etched with Nital 3%.

Related to the cange of its microstructure the heat treated material undergoes changes in mechanical properties in term of hardness. The initial hardness of the material on the thick part prior to the heat treatment was 165 HV by a test load of 1 kg. The thin impeller blade portion shows a hardness value of up to 705 HV, which decreases up to 115 HV after completing the heat treatment process.

## 3.3. Analysis

By heating at the upper temperature limit carbide in the ledeburite decomposes into graphite and ferrite. The total decomposition is achieved due to the sufficient 2 hour holding time, the small wall thickness 149 150 and the availability of silicon in the matrix which is also supported by relatively low levels of Manganese . Longer holding time results in not only the breakdown of carbides but also the diffusion of carbon out of the pearlite matrix. The iron carbide in the matrix of pearlite also breaks down into ferrite and graphite. During the decomposition process, the basic form of pearlite in carbide that resembles a spherical shape does not change. Graphite that is formed from carbon diffusion has a random and discrete orientation. Finally, graphites resemble as D-type graphite which is commonly derived from a eutectic reaction and associated with undercooling. The decrease of hardness occurs at the tip of the impeller drastically. This pattern of graphite deployment will result in a decrease in the value of tensile strength and impact. The remaining carbides can still be found because the initial setting of temperature calculations based on the remaining carbide is 5%.

4. Conclusion

😋 grammarly

The fully <u>ledeburitic</u> microstructure can <u>be transformed</u> into stable cast iron by applying heat treatment

process. To determined the heating temperature, a number of cementite in the matrix is put as the basic value for the calculation of carbon content while using the binary diagram of Fe-Fe3C. Ledeburite decomposes into ferrite and graphite. However the decomposition of ledeburite does not cause any increase of carbon content in the matrix, so that ferrite is still dominant in the matrix <sup>167</sup>. The process results in a structure of D-type graphite without any change in the shape and dimension of the previous primary phase. <sup>169</sup> The hardness decrease up to 115HV. Furthermore, the presence of Si and the low level of manganese are the requisites for the effective decomposition of cementite in the ledeburite.

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172,173

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Acknowledgments

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<mark>doi</mark> → DOI	Misspelled Words	Correctness
ledeburite	Unknown Words	Correctness
The water	Determiner Use (a/an/the/this, etc.)	Correctness
is made	Passive Voice Misuse	Clarity
ledeburite	Unknown Words	Correctness
<del>, This</del> → ; This, , and This, . This	Punctuation in Compound/Complex Sentences	Correctness
<del>ledeburite</del> → Lefebure	Misspelled Words	Correctness
ledeburite	Unknown Words	Correctness
, and	Punctuation in Compound/Complex Sentences	Correctness
<mark>sulphur</mark> → sulfur	Mixed Dialects of English	Correctness
<del>tablo</del> → Table	Misspelled Words	Correctness
<del>cost effective</del> → cost-effective	Misspelled Words	Correctness
<del>offective</del> → useful, active	Word Choice	Engagement
<mark>as is</mark> → as is	Improper Formatting	Correctness
<mark>good</mark> → excellent	Word Choice	Engagement
The impeller is made of gray cast iron with a material composition based on Fe with carbon, silicon, manganese alloy and some accompanying elements such as phosphorus and sulphur (table 1), Grey Cast Iron belongs to the most cost effective material for the impeller and classified as is one of the	Hard-to-read text	Clarity



<del>the matrix of</del>	Wordy Sentences	Clarity
be used	Passive Voice Misuse	Clarity
<del>licence</del> → license	Mixed Dialects of English	Correctness
and	Comma Misuse within Clauses	Correctness
<mark>licence</mark> → license	Mixed Dialects of English	Correctness
<del>loi</del> → DOI	Misspelled Words	Correctness
<del>affect</del> → Affect	Improper Formatting	Correctness
and	Punctuation in Compound/Complex Sentences	Correctness
he variation, or a variation	Determiner Use (a/an/the/this, etc.)	Correctness
a very	Determiner Use (a/an/the/this, etc.)	Correctness
<del>very thin</del> → fragile, skinny	Word Choice	Engagement
ledeburite	Unknown Words	Correctness
In addition → Also, Besides	Wordy Sentences	Clarity
the occurrence of	Wordy Sentences	Clarity
ledeburite	Unknown Words	Correctness
be initiated	Passive Voice Misuse	Clarity
, and	Comma Misuse within Clauses	Correctness
ledeburite	Unknown Words	Correctness
<del>, however</del> → . However, ; however	Punctuation in	Correctness



	Compound/Complex Sentences	
be formed	Passive Voice Misuse	Clarity
<del>, whereas it</del> → . In contrast, it	Hard-to-read text	Clarity
<mark>larger</mark> → more significant	Word Choice	Engageme
a bigger	Determiner Use (a/an/the/this, etc.)	Correctne
	Intricate Text	Clarity
ledeburite	Unknown Words	Correctne
The solidification of metal liquid in the area of smaller wall thickness will form a ledeburite microstructure.	Hard-to-read text	Clarity
<del>to be</del>	Wordy Sentences	Clarity
ledeburite	Unknown Words	Correctne
<del>doi</del> → DOI	Misspelled Words	Correctne
ledeburite	Unknown Words	Correctne
ledeburite	Unknown Words	Correctne
is accomplished	Passive Voice Misuse	Clarity
austenisation → austenitization, customisation	Misspelled Words	Correctne
The cooling	Determiner Use (a/an/the/this, etc.)	Correctne
be set	Passive Voice Misuse	Clarity
the CCT	Determiner Use (a/an/the/this,	Correctne



53.	ledeburite	Unknown Words	Correctness
54.	<del>apropriate</del> → appropriate	Misspelled Words	Correctness
55.	ledeburite	Unknown Words	Correctness
56.	be transformed	Passive Voice Misuse	Clarity
57.	is dispersed	Passive Voice Misuse	Clarity
58.	initially presented, presented initially	Word Choice	Engagement
59.	heattreatment → heat treatment, heat-treatment	Misspelled Words	Correctness
60.	ledeburite	Unknown Words	Correctness
61.	, such	Punctuation in Compound/Complex Sentences	Correctness
62.	manganese; Manganese	Text Inconsistencies	Correctness
63.	<del>of nodular</del> → of nodular	Improper Formatting	Correctness
64.	<del>down the</del> → down the	Improper Formatting	Correctness
65.	<mark>iron</mark> → metal	Word Choice	Engagement
66.	ledeburite	Unknown Words	Correctness
67.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
68.	<mark>Sampel</mark> → Sample	Misspelled Words	Correctness
69.	ledeburite	Unknown Words	Correctness
70.	, which	Punctuation in Compound/Complex Sentences	Correctness
71.	ledeburite	Unknown Words	Correctness



72.	micro structure → microstructure	Confused Words	Correctness
73.	<del>of the</del> → of the	Improper Formatting	Correctness
74.	be used	Passive Voice Misuse	Clarity
75.	matrix after → matrix after	Improper Formatting	Correctness
76.	<del>after the</del> → after the	Improper Formatting	Correctness
77.	ledeburite	Unknown Words	Correctness
78.	is set	Passive Voice Misuse	Clarity
79.	the y	Determiner Use (a/an/the/this, etc.)	Correctness
80.	be applied	Passive Voice Misuse	Clarity
81.	ledeburite	Unknown Words	Correctness
82.	ledeburite	Unknown Words	Correctness
83.	ledeburite	Unknown Words	Correctness
84.	Microstructure Carbon content Weight percent	Intricate Text	Clarity
85.	, it	Punctuation in Compound/Complex Sentences	Correctness
86.	ledeburite	Unknown Words	Correctness
87.		Intricate Text	Clarity
88.	ledeburite	Unknown Words	Correctness
89.	Microstructure Max.	Incomplete Sentences	Correctness
90.	is accordingly based	Passive Voice Misuse	Clarity
91.	is used	Passive Voice Misuse	Clarity



92.	heating,	Punctuation in Compound/Complex Sentences	Correctness
93.	<mark>shown</mark> → is shown	Incorrect Verb Forms	Correctness
94.		Intricate Text	Clarity
95.	<del>this</del> Figure	Determiner Use (a/an/the/this, etc.)	Correctness
96.	the minimum	Determiner Use (a/an/the/this, etc.)	Correctness
97.	, and	Punctuation in Compound/Complex Sentences	Correctness
98.	, and	Punctuation in Compound/Complex Sentences	Correctness
99.	the solidus	Determiner Use (a/an/the/this, etc.)	Correctness
100.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
101.	ledeburite	Unknown Words	Correctness
102.	exactly → precisely	Word Choice	Engagement
103.	, which	Punctuation in Compound/Complex Sentences	Correctness
104.	The heating time is set to 2 hours according to the minimum reaction time to disolve the ledeburite for a thickness of 5mm.	Wordy Sentences	Clarity
105.	<mark>disolve</mark> → dissolve	Misspelled Words	Correctness
106.	ledeburite	Unknown Words	Correctness
107.	<mark>ledeburite</mark> → Lefebure	Misspelled Words	Correctness



108.	were performed	Passive Voice Misuse	Clarity
109.	<del>vickers</del> → Vickers	Misspelled Words	Correctness
110.	is formed	Passive Voice Misuse	Clarity
111.	<del>ledeburite</del> → Lefebure	Misspelled Words	Correctness
112.	, assemblies	Punctuation in Compound/Complex Sentences	Correctness
113.	doi → DOI	Misspelled Words	Correctness
114.	at	Wordy Sentences	Clarity
115.	a spherical	Determiner Use (a/an/the/this, etc.)	Correctness
116.	is already transformed	Passive Voice Misuse	Clarity
117.	<del>not</del> → no	Confused Words	Correctness
118.	<del>exist</del> → exists	Faulty Subject-Verb Agreement	Correctness
119.	is transformed	Passive Voice Misuse	Clarity
120.	which was	Wordy Sentences	Clarity
121.	originally → initially	Word Choice	Engagement
122.	was originally located	Passive Voice Misuse	Clarity
123.	, which	Punctuation in Compound/Complex Sentences	Correctness
124.	is commonly identified	Passive Voice Misuse	Clarity
125.	$-$ Conditions $\rightarrow$ —conditions	Incomplete Sentences	Correctness
126.	<del>tipe</del> → type, tipi	Misspelled Words	Correctness



127.	is considered	Passive Voice Misuse	Clarity
128.	Retained iron → Retained iron	Improper Formatting	Correctness
129.	i <del>ron carbide</del> → iron carbide	Improper Formatting	Correctness
130.	<mark>doi</mark> → DOI	Misspelled Words	Correctness
131.	<del>cange</del> → change	Misspelled Words	Correctness
132.	microstructure,	Punctuation in Compound/Complex Sentences	Correctness
133.	heat treated → heat-treated	Misspelled Words	Correctness
134.	<del>undergoes changes in</del> → changes	Wordy Sentences	Clarity
135.	<del>torm</del> → terms	Incorrect Noun Number	Correctness
136.	material → content	Word Choice	Engagement
137.	<del>prior to</del> → before	Wordy Sentences	Clarity
138.	<del>up</del>	Wordy Sentences	Clarity
139.	upper-temperature	Misspelled Words	Correctness
140.	, carbide	Punctuation in Compound/Complex Sentences	Correctness
141.	ledeburite	Unknown Words	Correctness
142.	is achieved	Passive Voice Misuse	Clarity
143.	, and	Comma Misuse within Clauses	Correctness
144.	<del>and the</del> → and the	Improper Formatting	Correctness
145.	of silicon → of silicon	Improper Formatting	Correctness

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<mark>silicon in</mark> → silicon in	Improper Formatting	Correctness	
<del>matrix which</del> → matrix which	Improper Formatting	Correctness	
, which	Punctuation in Compound/Complex Sentences	Correctness	
<mark>also supported</mark> → also supported	Improper Formatting	Correctness	
<del>supported by</del> → supported by	Improper Formatting	Correctness	
	Intricate Text	Clarity	
	Intricate Text	Clarity	
<del>matrix</del> → model, pattern, array	Word Choice	Engagement	
<del>basic</del> → primary	Word Choice	Engagement	
is formed	Passive Voice Misuse	Clarity	
<del>88</del>	Wrong or Missing Prepositions	Correctness	
which	Punctuation in Compound/Complex Sentences	Correctness	
<del>decrease in</del> → reduction of	Word Choice	Engagement	
be found	Passive Voice Misuse	Clarity	
ledeburitic	Unknown Words	Correctness	
be transformed	Passive Voice Misuse	Clarity	
<mark>a number of</mark> → several, some, many	Wordy Sentences	Clarity	
a number of cementite	Misuse of Quantifiers	Correctness	
<mark>basic</mark> → primary, underlying, fundamental, essential	Word Choice	Engagement	



However,	Punctuation in Compound/Complex Sentences	Correctness
ledeburite	Unknown Words	Correctness
matrix → form, model, pattern, array	Word Choice	Engagement
	Intricate Text	Clarity
	Intricate Text	Clarity
effective → active, useful	Word Choice	Engagement
ledeburite	Unknown Words	Correctness
Cost effective → Cost-effective	Misspelled Words	Correctness
effective → useful, active	Word Choice	Engagement
<mark>and</mark> → And	Improper Formatting	Correctness
<mark>doi</mark> → DOI	Misspelled Words	Correctness
was conducted	Passive Voice Misuse	Clarity
POLMAN → Polman	Confused Words	Correctness
you for	Pronoun Use	Correctness
support.	Closing Punctuation	Correctness
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